

WEEKLY INSTRUCTIONAL PLAN

TEACHER: Thomas		WEEK OF: 10/17/22 - 10/21/22 (B week)
MONDAY	SUBJECT: Chemistry (1st and 8th) CLASSWORK: <ol style="list-style-type: none"> 1. DN - AI question 2. Democritus 3. Mystery Boxes 4. Timeline template in notebook and into to scientist project 5. Exit - why? HOMEWORK: none	SUBJECT: DC Chemistry CLASSWORK: <ol style="list-style-type: none"> 1. DN - what is a mole? (Aktiv Chem practice) 2. molar mass and conversions (practice) 3. formula mass (practice) 4. Lab 5 - Day 3 5. Exit - lab conclusions HOMEWORK: lab write-up
TUESDAY	SUBJECT: Chemistry CLASSWORK: <ol style="list-style-type: none"> 1. DN - AI question 2. Democritus 3. Mystery Boxes 4. Timeline template in notebook and into to scientist project 5. Dalton and Atomic Theory 6. JJ Thomson 7. Exit - how do you know? HOMEWORK: none	SUBJECT: DC Chemistry CLASSWORK: NA HOMEWORK: see above
WEDNESDAY	SUBJECT: Chemistry (1st and 8th)	SUBJECT: DC Chemistry

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	<p>CLASSWORK:</p> <ol style="list-style-type: none"> 1. DN - compare Atomic Theory to JJ Thomson's atom 2. Rutherford 3. subatomic particle chart 4. vocabulary 5. exit - card match <p>HOMEWORK: none</p>	<p>CLASSWORK:</p> <ol style="list-style-type: none"> 1. DN - molar mass problems 2. Lab 5 calculations 3. % composition 4. empirical formulas 5. strip problems 6. Exit - demo problem <p>HOMEWORK: mole problems</p>
THURSDAY	<p>SUBJECT: Chemistry</p> <p>CLASSWORK:</p> <ol style="list-style-type: none"> 1. DN - compare Atomic Theory to JJ Thomson's atom 2. Rutherford 3. subatomic particle chart 4. vocabulary 5. Isotopes - example and practice 6. Exit - ck12 quiz <p>HOMEWORK: none</p>	<p>SUBJECT: DC Chemistry</p> <p>CLASSWORK: NA</p> <p>HOMEWORK: NA</p>
FRIDAY	<p>SUBJECT: Chemistry</p> <p>CLASSWORK: complete above (8th pd - pep rally???)</p> <p>HOMEWORK: none</p>	<p>SUBJECT: DC Chemistry</p> <p>CLASSWORK:</p> <ol style="list-style-type: none"> 1. DN - empirical formula practice (Aktiv Chem) 2. Mole Day activities (mole lab tasks) 3. molarity/concentration

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		<p>4. exit - make 50 mL of a 0.5 M solution of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$</p> <p>HOMEWORK: lab? or OpenStax reading/practice</p>
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